## DR-60. THE RESEARCH ON PHASE COMPOSITION AND POROUS STRUCTURE OF THE INDIUM SULPHIDE (III) NANOMATERIALS OBTAINED VIA CHEMICAL BATH DEPOSITION

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At present thin films of indium sulphide  $In_2S_3$  are promising materials for solar energy. The aim of this work was to study the properties of  $In_2S_3$  based layers and powders obtained by the chemical bath deposition method [1].

A set of the samples of indium sulphide (III) was synthesized using the solutions with various composition. The measurements of the specific surface area of  $\rm In_2S_3$  nanopowders were carried out by nitrogen thermal desorption method in the SORBI MS device [2]. The morphology of the surface layers was studied with the use of atomic force microscope (AFM) Ntegra Terma. To study the phase composition of the layers, Raman spectroscopy (LabRam HR800 spectrometer) was used.

As a result of chemical synthesis, thin layers were obtained with a thickness of 300 to 850 nm, depending on the conditions of deposition. Using the method of nitrogen desorption, the specific surface area per mass unit for each of the powders from the series of samples was determined. This information allowed us to estimate the average particle size, which ranges from 40 to 90 nm. The decrease in particle sizes is due to a change in the concentration of the metal salt in the initial solution. With an increase in the metal content in the initial solution, the roughness of the films also decreased, which was established when the samples were analyzed by the AFM method in a semi-contact mode.

In the study of the samples by Raman spectroscopy, blurred Raman spectra of the samples were recorded, which indicates a small crystallization of the structure of the films under study. However, the position of the reflections corresponds to the crystalline phase of  $\rm In_2S_3$  [3]. When conducting thermal annealing in the temperature range from 50 °C to 350 °C, new phases are not observed, the degree of crystallization of the structure is preserved, which indicates the thermal stability of the samples under study in this temperature range.

## References

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