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# Homogeneity, metallurgical, mechanical, wear, and corrosion behavior of Ni and B<sub>4</sub>C coatings deposited on 304 stainless steels developed by microwave cladding technique

Shashi Prakash Dwivedi <sup>a,\*\*</sup>, Shubham Sharma <sup>b,c,d,\*</sup>, Arun Pratap Srivastava <sup>a</sup>, Vandana Arora Sethi <sup>e</sup>, Kahtan A. Mohammed <sup>f,g</sup>, Abhinav Kumar <sup>h</sup>, M. Ijaz Khan <sup>d,i</sup>, Mohamed Abbas <sup>J</sup>, Elsayed M. Tag-Eldin <sup>k,\*\*\*</sup>

- <sup>a</sup> Lloyd Institute of Engineering & Technology, Plot No. 3, Knowledge Park II, Greater Noida, Uttar Pradesh 201306, India
- b Mechanical Engineering Department, University Centre for Research and Development, Chandigarh University, Mohali, Punjab, 140413, India
- <sup>c</sup> School of Mechanical and Automotive Engineering, Qingdao University of Technology, 266520, Qingdao, China
- d Department of Mechanical Engineering, Lebanese American University, Kraytem 1102-2801, Beirut, Lebanon
- <sup>e</sup> Lloyd Institute of Management and Technology, Plot No 11, Knowledge Park II, Greater Noida, Uttar Pradesh 201306, India
- f Faculty of Pharmacy, Jabir Ibn Hayyan Medical University, Najaf, Iraq
- g Department of Medical Physics, Hilla University College, Babylon, Iraq
- h Department of Nuclear and Renewable Energy, Ural Federal University Named After the First President of Russia, Boris Yeltsin, 19 Mira Street, 620002, Ekaterinburg, Russia
- i Department of Mechanics and Engineering Science, Peking University, Beijing 100871, China
- <sup>j</sup> Electrical Engineering Department, College of Engineering, King Khalid University, Abha 61421, Saudi Arabia
- k Faculty of Engineering, Future University in Egypt, 11835 Cairo, Egypt

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#### ABSTRACT

The microwave cladding technique for depositing Ni and 10 %  $B_4C$  coatings on 304 stainless steel has yielded significant advancements in material properties and performance. The key findings of this study revealed remarkable improvements, including a 43.33% increase in material hardness, indicating enhanced wear resistance and mechanical properties. This improvement was attributed to the uniform distribution of  $B_4C$  and Ni on the cladding surface, ensuring a consistent interfacial layer developed between SS 304 and the cladding surface without cracks and porosity. Microstructural analysis at  $500 \times$  magnification unveiled an impressive 2233.35 grains per square inch, showcasing the refined grain structure achieved during the cladding process. Wear testing demonstrated a low wear rate of  $0.00308 \text{ mm}^3/\text{m}$  and a favorable coefficient of friction of 0.1981, confirming the material's suitability for applications with demanding frictional conditions. Furthermore, the corrosion behavior of the coated 304 stainless steel was assessed, revealing a minimal corrosion weight loss of only 0.42 mg for the Ni and 10%  $B_4C$  coated sample. The presence of various carbide phases, such as  $Cr_2C$ ,  $Cr_2G_6$ ,  $Cr_7BC_4$ ,  $Fe_5C_2$ , and  $Fe_{23}B_6$ , within the cladding further contributed to the material's enhanced mechanical and wear properties.

#### 1. Introduction

Steel is one of the most widely used materials in the world due to its exceptional strength and durability. Steel is a ubiquitous material in modern industry, and can be significantly enhanced through the

incorporation of various ceramic particles as coatings [1]. This innovative approach combines the strength and malleability of steel with the exceptional properties of ceramics, opening up a wide range of applications across different sectors [2,3]. One common ceramic material used for steel coating is alumina (Al<sub>2</sub>O3). Alumina coatings are prized

E-mail addresses: spdglb@gmail.com (S.P. Dwivedi), shubham543sharma@gmail.com, shubhamsharmacsirclri@gmail.com (S. Sharma), Elsayed.tageldin@fue.edu.eg (E.M. Tag-Eldin).

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<sup>\*</sup> Corresponding author. Lloyd Institute of Engineering & Technology, Plot No. 3, Knowledge Park II, Greater Noida, Uttar Pradesh 201306, India.

<sup>\*\*</sup> Corresponding author.

<sup>\*\*\*</sup> Corresponding author.

for their excellent hardness, wear resistance, and corrosion resistance [4]. When applied to steel surfaces, they form a protective layer that shields against abrasive wear, extending the lifespan of components in heavy machinery and cutting tools. Another notable ceramic coating is zirconia (ZrO2). Zirconia-coated steel exhibits superior thermal and chemical resistance, making it suitable for applications in high-temperature environments such as gas turbines and industrial furnaces [5,6]. Zirconia coatings can also enhance the electrical insulation properties of steel, a valuable feature for electronic and electrical applications. Additionally, silicon carbide (SiC) coatings are known for their extreme hardness and abrasion resistance [7,8]. When combined with steel, SiC coatings find applications in armor protection, cutting tools, and high-performance bearings [9-12]. Titanium nitride (TiN) coatings are popular for their gold-like appearance and excellent wear resistance. They are often used on steel tools and components in the automotive and aerospace industries [13-15].

Boron carbide ( $B_4C$ ) is a versatile material used in various industrial applications, including its utilization as a coating for steel surfaces [16]. This ceramic compound offers several advantages in enhancing the performance and durability of steel components. First and foremost,  $B_4C$  coatings are renowned for their exceptional hardness [17]. This property makes them highly resistant to abrasion, wear, and scratches, which is particularly beneficial for steel used in harsh environments or subjected to heavy mechanical stress [18–20]. Moreover, B4C coatings provide excellent corrosion resistance. Steel is susceptible to rust and degradation when exposed to moisture and corrosive substances, but a  $B_4C$  coating acts as a protective barrier, extending the lifespan of the steel component [21–23]. Additionally,  $B_4C$ -coated steel can exhibit improved thermal stability and electrical conductivity, making it suitable for applications in high-temperature environments and electronic industries [24–26].

The combination of nickel (Ni) with boron carbide (B<sub>4</sub>C) for steel coating offers a unique set of advantages in terms of enhancing the performance and properties of steel components [27,28]. This composite coating is particularly valuable in demanding industrial applications. Nickel, known for its excellent corrosion resistance and electrical conductivity, provides a robust protective layer when combined with B<sub>4</sub>C. B<sub>4</sub>C is exceptionally hard and wear-resistant, making it ideal for guarding steel against abrasive wear and scratches [29,30]. The synergy between Ni and B<sub>4</sub>C in a coating improves the overall hardness, wear resistance, and corrosion resistance of the steel substrate [31,32]. This, in turn, extends the lifespan of steel components, reduces maintenance costs, and enhances their performance in harsh environments. Additionally, the electrical conductivity of nickel allows for specific applications where electrical or thermal conductivity is crucial.

Traditional cladding techniques, although effective in many applications, come with several inherent problems that have driven the exploration of alternative methods like microwave cladding. Traditional cladding methods like welding or thermal spraying can lead to variations in coating thickness, resulting in uneven protection and performance [33,34]. Welding processes often introduce HAZ, altering the mechanical properties of the substrate and creating potential weak points. Some cladding materials are incompatible with certain substrates, limiting material choices and applications. Conventional cladding methods can be energy-intensive, leading to higher operational costs and environmental concerns. Thermal spraying and similar methods may have relatively slow deposition rates, impacting productivity. Material waste is common in traditional cladding processes due to overspray, spatter, or excessive heat [35,36]. Microwave cladding offers a potential solution to the traditional cladding problems. Microwave cladding allows for precise control of temperature distribution, leading to uniform coating thickness and properties across the substrate. Microwave cladding operates at lower temperatures, reducing the likelihood of heat-affected zones and substrate property alterations. Microwave cladding can be applied to a wide range of materials and substrates, enhancing compatibility and expanding applications.

Microwave heating is efficient and can significantly reduce energy consumption compared to traditional methods. Microwave cladding can achieve rapid heating and deposition, increasing production rates [37]. With precise material distribution, microwave cladding can minimize waste, making it a more cost-effective and environmentally friendly option.

In a study by B. Singh et al. [38], the microwave cladding behavior was reviewed, and it was strongly recommended to employ protective coatings to prevent erosion damage. C. Durga Prasad et al. [39] innovatively created a cladding surface on AISI 410 steel through microwave irradiation, using a combination of CoMoCrSi and Flyash. X-ray diffraction (XRD) analysis unveiled the presence of various hard phases, including Mo<sub>3</sub>Si, SiC, TiC, Co<sub>3</sub>Ti, and Cr<sub>3</sub>C<sub>2</sub>. This research demonstrates the effectiveness of microwave-based methods in producing enhanced surface coatings with a combination of hard phases for improved material properties. In their work, B. Singh et al. [40] employed a modified B-type thermocouple to create a Ni-based cladding while simultaneously measuring real-time temperature distribution. This innovative approach allowed for precise monitoring of temperature changes during the cladding process, facilitating better control and understanding of the thermal dynamics involved.

J.S. Vishwanatha et al. [41] successfully developed microwave clads on an AISI 420 substrate within a remarkably short exposure time of just 7 min. Their research revealed significant improvements in erosion resistance for the clad surfaces, surpassing the performance of the original substrates. This achievement highlights the efficacy of microwave cladding in enhancing the material's resistance to erosion, which has implications for various industrial applications requiring improved durability. K. P. Kumar et al. [42] employed nickel microwave cladding to significantly enhance the surface properties of austenitic stainless steel. The cladding process resulted in improved resistance to oxidation and wear, though a thin oxide layer was observed on clad surfaces. Notably, the clad region exhibited a substantially higher microhardness of 458 HV compared to the substrate region's 234 HV. The presence of silicon, cobalt, and chromium in the cladding further contributed to the material's enhanced surface characteristics.

Bhupinder Singh et al. [43] investigated the influence of post-clad heat treatment on Ni-based microwave cladding. Their findings revealed that at 1000 °C, the clad material undergoes partial dissolution of eutectic carbides during heat treatment, shedding light on the thermal transformations occurring within the cladded structure. T. K. Mishra et al. [44] employed microwave hybrid heating to deposit nickel and nickel-tungsten carbide cladding on a low-carbon steel base. The research highlighted that incorporating 12 wt % of tungsten carbide (WC) significantly improved the cladding's resistance to abrasive wear, microstructure, and hardness. Interestingly, the Ni-12 wt% WC coating exhibited greater hardness compared to the Ni-24 wt% WC coating, suggesting an optimal composition for enhanced material properties. A. Babu et al. [45] synthesized Ni-SiC composite claddings, revealing remarkable corrosion resistance, with Ni-SiC bimodal cladding outperforming WCCoCr coating by sevenfold. However, Ni-SiC microwave claddings exhibited a combination of brittle and ductile degradation processes, underscoring the importance of considering both their strengths and limitations in various applications. Rakesh. B. Nair et al. [46] utilized microwave heating to develop AlxCoCrFeNi high entropy alloy claddings. These claddings exhibited impressive erosion resistance, with the equimolar high entropy alloy showing erosion rates nearly 23 times lower than those observed in SS316L steel, emphasizing their potential as superior materials for applications where erosion resistance is critical. Sunny Zafar et al. [47] harnessed microwave radiation to fabricate clads featuring nanostructures, with nano-carbides distributed in a consistent pattern. The produced clads exhibited impressive mechanical properties, boasting a flexural strength averaging  $671 \pm 28$  MPa and a microhardness of 1760  $\pm$  128 HV. This highlights the potential of microwave cladding for engineering materials with enhanced nanostructural features and remarkable mechanical strength.

Ajit M. Hebbale et al. [48] employed microwave irradiation to create a Ni-based coating on austenitic SS-304. Microstructure analysis revealed a remarkably low porosity of approximately 0.87%, while increased hardness was attributed to the dispersion of intermetallics and metal carbides. The resulting clad surface exhibited a notable microhardness, averaging  $364 \pm 70$  HV, showcasing the efficacy of microwave cladding for enhancing material properties and surface hardness.

The research gap in the present study lies in the limited understanding of the comprehensive and synergistic impact of microwave cladding technology on the properties of Ni and  $B_4C$  coatings when applied to 304 stainless steels. Despite significant advancements in coating technologies and their applications in various industries, there is a distinct lack of research that investigates the combined effect of microwave cladding, Ni coatings, and  $B_4C$  in the context of 304 stainless steel. Existing studies often focus on individual aspects, such as coating material or deposition technique, but fail to address the holistic evaluation of these multifaceted coatings on the specific substrate. This research gap calls for a thorough investigation to bridge the knowledge void and provide valuable insights into the performance of these coatings in a real-world application.

This research aims to address a critical knowledge gap by pioneering a comprehensive study of the homogeneity, metallurgical attributes, mechanical properties, and the wear and corrosion behavior of Ni and B4C coatings when applied to 304 stainless steels using the innovative microwave cladding technique. The novelty of this study lies in its holistic approach, as it integrates multiple factors that have not been studied collectively before. By using the novel microwave cladding method to deposit Ni and B<sub>4</sub>C coatings on 304 stainless steels, this research will provide unique insights into the synergistic effects of these coatings, offering a significant advancement in the field of materials science and surface engineering. The objective of this study is to develop and characterize Ni and B<sub>4</sub>C-coated SS-304 steel using the microwave cladding technique. The specific aims of the study are to investigate the microstructural and mechanical properties of the coated steel and evaluate the coating's resistance to wear, and corrosion.

# 2. Materials and methods

# 2.1. Base material

SS 304 has a face-centered cubic (FCC) crystal structure, which contributes to its excellent mechanical and corrosion-resistant properties. The FCC structure allows for close-packed atomic arrangements, enhancing its strength and ductility. SS 304 has a density of around 8 g/cm³, making it moderately dense. Thermal Conductivity: It has a relatively low thermal conductivity of about 16.2 W/(m·K). SS 304 exhibits moderate electrical conductivity. SS 304 is primarily composed of iron, chromium (18–20%), and nickel (8–10.5%). It contains small amounts of carbon ( $\leq$ 0.08%) and manganese ( $\leq$ 2%), which contribute to its corrosion resistance and weldability. The presence of chromium forms a passive oxide layer on the surface, providing excellent corrosion resistance, especially in various corrosive environments [1]. Table 1 displays analyzed the properties of SS-304 up to 160 mm diameter/thickness.

## 2.2. Primary cladding particle

Fig. 1 shows powder XRD of Ni powder used in the present study. The  $\,$ 

Table 1

Analyzed the Properties of SS-304 up to 160 mm diameter/thickness [49].

S. No.	Properties	Values
1	Density (g/cm <sup>3</sup> )	8
2	Tensile Strength (MPa)	615
3	Melting Temperature (Degree centigrade)	1450
4	Vicker Hardness (HV)	210

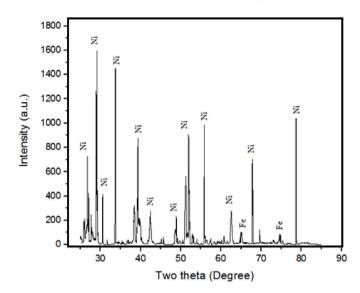


Fig. 1. Powder XRD of Ni powder.

average particle size of Ni was 18  $\mu$ m. The application of nickel (Ni) coating on steel substrates, combined with boron carbide (B<sub>4</sub>C) reinforcement, using microwave cladding techniques, offers significant enhancements to the material's properties. Ni coating provides several advantages, such as corrosion resistance, increased hardness, and enhanced wear resistance. When B<sub>4</sub>C particles are incorporated into the Ni matrix, the resulting composite exhibits exceptional properties, including improved strength and abrasion resistance [27]. The microwave cladding technique allows for a homogenous distribution of B<sub>4</sub>C particles within the Ni coating, ensuring superior mechanical and thermal properties. The coating thickness range of Ni with B<sub>4</sub>C on steel can vary depending on the specific application and the desired performance characteristics [29]. However, for better surface properties, it is essential to consider a coating thickness range that balances both functionality and cost-effectiveness. Here are some general guidelines.

# 2.2.1. Thin coatings (5–50 μm)

- Thin coatings are suitable for applications where surface enhancement is the primary goal, such as improving wear resistance, corrosion resistance, or surface hardness.
- These coatings are typically cost-effective and may provide good results for applications with moderate wear or corrosion challenges.

# 2.2.2. Moderate coatings (50–200 $\mu m$ )

- Moderate coatings strike a balance between cost and performance.
- They are suitable for applications where higher wear resistance or improved surface properties are required, but there are constraints on coating thickness.

# 2.2.3. Thick coatings (>200 $\mu$ m)

- Thick coatings are used when extreme wear resistance, extended service life, or severe environmental conditions are expected.
- These coatings may be necessary for heavy-duty industrial equipment or components subjected to aggressive wear and corrosion.

# 2.3. Secondary cladding particle

Boron carbide (B<sub>4</sub>C) with an average particle size of 20  $\mu$ m was chosen as a secondary reinforcement particle. B<sub>4</sub>C is manufactured through various methods, including solid-state reaction processes and

chemical vapor deposition (CVD). It is often synthesized from a mixture of boron oxide and carbon-containing precursors at high temperatures.  $B_4C$  has a complex crystal structure, primarily consisting of boron (B) and carbon (C) atoms arranged in an icosahedral boron-based lattice. This structure imparts  $B_4C$  with remarkable hardness, making it one of the hardest materials known, second only to diamond.  $B_4C$  typically consists of approximately 77–80% boron and 20–23% carbon. Its chemical composition makes it resistant to chemical reactions with most acids and bases.  $B_4C$  is renowned for its exceptional hardness, high modulus of elasticity, and excellent wear resistance [31]. Fig. 2 is an X-ray powder diffraction pattern of the  $B_4C$  particles employed in this investigation. The X-ray powder diffraction pattern of  $B_4C$  powder reveals  $B_4C$ ,  $B_2O_3$ , and C phases as illustrated in the Fig. 3.

# 2.4. Development of cladding

In the present study, an In-house developed microwave oven was used for the cladding process. Microwave hybrid heating was carried out at 900 W with 2.45 GHz frequency (Table 2). Alcohol was used in an ultrasonic bath to clean the SS-304 substrate before deposition. Ni and B<sub>4</sub>C particles (10% of the total) were heat treated in a muffle furnace for 24 h at 1100 °C. Ni and 10 % B<sub>4</sub>C combination was preheated to drive out any remaining moisture. The warmed powder was sprinkled on the SS-304 substrate in a roughly even layer. Microwave interaction is material-dependent, with a key factor being skin depth, particularly crucial for swift interaction. In the context of hard-facing powder, maintaining the primary ingredient at a 4.5 m skin depth at 2.45 GHz helps avoid direct particle interaction with microwave radiation at ambient temperature, optimizing the cladding process. To mitigate microwave reflection issues caused by a Ni and 10% B4C powder blend clads were fabricated using charcoal as the susceptor material via microwave hybrid heating (MHH). This substitution effectively addresses reflection challenges, ensuring efficient heating and material processing in the MHH method [47,48]. At room temperature, charcoal exhibited strong microwave interaction, resulting in rapid heating. This phenomenon subsequently elevated the temperature of the Ni and 10% B4C powder in the mixture, facilitating effective and efficient processing in the desired application. The experimental conditions for creating the microwave-clad samples from a 10% B<sub>4</sub>C and Ni powder on the SS-304 mixture are shown in Table 2. To prevent contamination of the clad, a pure graphite sheet was interposed between the susceptor and the Ni and 10% B4C powder. The MHH process was executed at 900 W and 2.45 GHz using a multimode microwave applicator, ensuring controlled

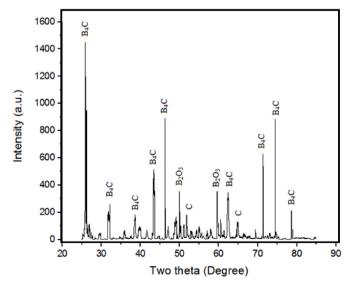


Fig. 2. Powder XRD of B<sub>4</sub>C powder.

and efficient heating while safeguarding the quality of the final product. The powder layer underwent 120 s of irradiation. Microwave-clad samples were fabricated using a blend of 10% B4C and Ni powder on SS-304, demonstrating the effectiveness of the microwave cladding process for enhancing the material's properties. A uniform cladding surface was obtained by controlling the cladding process parameters. Optimum process parameters were decided to control the uniform cladding surfaces as per the pilot run investigation as shown in Table 2. Randomly, Irradiation time was chosen 100 s, and other parameters were kept constant. The presence of weak spots and gaps between the substrate and cladding surface was detected with 100 s Irradiation time. This suggested that the cladding process had encountered challenges related to material bonding or uniform deposition within the given timeframe. An issue of porosity became evident on the cladding surface during a 150-s microwave cladding process. This occurrence was attributed to the prolonged irradiation, causing excessive heating of the substrate material. The elevated substrate temperature led to increased gas entrapment within the molten cladding material, resulting in porosity. During a 120-s irradiation time in a cladding process, a significant achievement was the attainment of a uniform cladding surface that was entirely free from porosity. This outcome signified the successful control of irradiation time to ensure the even distribution and bonding of the cladding material onto the substrate. The absence of porosity indicated minimal gas entrapment or voids within the cladding layer, enhancing its structural integrity and corrosion resistance. Following the pilot run investigation, an irradiation time of 120 s was deemed optimal for subsequent cladding processes, ensuring the desired results. The same course of action was conducted to decide the other process parameters (Table 2).

#### 2.5. Materials testing and equipment

To conduct microstructure analysis, X-ray diffraction (XRD), and hardness measurements on cladding samples of SS 304, you will need specific testing equipment with the following properties [50–55].

#### 2.5.1. Scanning electron microscope (SEM)

- Type: Field-emission SEM (FE-SEM) is used for high-resolution imaging.
- Detector: Secondary Electron (SE) and Backscattered Electron (BSE) detectors for surface imaging and compositional contrast were used.
- Sample stage: Motorized stage for precise sample positioning.
- Chamber: A vacuum environment was used to minimize electron scattering.

# 2.5.2. X-ray diffraction (XRD) equipment

- XRD type: A powder X-ray diffractometer was used for phase analysis.
- Detector: A high-resolution, sensitive detector such as a scintillation
  detector.
- Goniometer: Motorized goniometer for precise sample positioning.
- Sample holder: A sample holder that accommodates cladding samples and enables rotation.

# 2.5.3. Hardness testing equipment

- Hardness tester: Vickers hardness tester on the hardness scale.
- Load application: Motorized load application system for accurate and consistent loads.

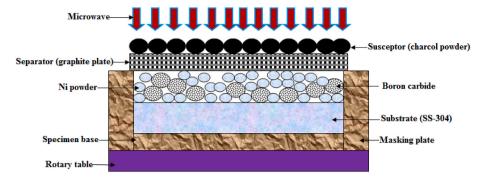


Fig. 3. Experimental procedure.

**Table 2**Microwave cladding process parameters [49].

S. No.	Parameters	Description
1	Frequency	2.45 GHz
2	Power	900 W
3	Susceptor	Charcoal
4	Separator	Graphite plate
5	Preheating powder Temperature	120 °C
6	Preheating time of powder	22 h
7	Average cooling time	680 s
8	Irradiation time	120 s
9	Atmosphere	Ambient
10	Powder	$Ni+10\ \%\ B_4C$
11	Substrate	SS-304

#### 2.6. Wear testing

The wear testing procedure on SS 304 steel coated with Ni and 10% B<sub>4</sub>C using a pin-on-disc apparatus typically follows these steps [56–58].

# 2.6.1. Specimen preparation

 $\bullet\,$  The SS 304 specimen was prepared with the Ni and 10%  $B_4C$  coating.

#### 2.6.2. Pin-on-disc setup

 The pin-on-disc test apparatus was set up, consisting of a rotating disc and a fixed pin (counterpart). The disc was made of a material representative of the wear conditions.

# 2.6.3. Load and speed configuration

- A specified load was applied (5 N) on the pin.
- The rotational speed of the disc was set (2 m/s).

# 2.6.4. Wear test run

- The coated SS 304 specimen (pin) was placed in contact with the rotating disc.
- The pin-on-disc apparatus was run for a predetermined distance or time (1000 m).

#### 2.6.5. Data collection

• The weight loss of the pin (coated SS 304 specimen) was measured and recorded before and after the test to determine the wear rate.

#### 2.6.6. Friction measurement

• The coefficient of friction during the test was monitored and recorded to assess the material's tribological behavior as exhibited in the Fig. 4.

# 3. Results and discussions

# 3.1. Microstructure investigation

Optical micrographs of clad SS-304 alloy with a 10% B<sub>4</sub>C and Ni powder microwave coating are displayed in Fig. 5. Surface areas of SS-304 alloy were able to dissolve Ni powder. However, the steel's surface shows signs of having been contaminated with B<sub>4</sub>C powder. In addition to improving the surface quality of steel, Ni powder was also a suitable coating material in and of itself. Introducing a combination of 10% B<sub>4</sub>C and nickel (Ni) to steel has yielded remarkable enhancements in surface properties. The most notable improvements lie in heightened hardness and wear resistance. This was largely attributed to the achievement of a uniform distribution of B<sub>4</sub>C and Ni (Fig. 6), which resulted in a surface entirely free from porosity and cracks. Such a surface transformation was pivotal in extending the steel's longevity and ensuring it performed exceptionally well in environments characterized by abrasion and wear. The synergy between B<sub>4</sub>C and Ni presented a significant advancement in fortifying the steel's surface. However, the structure of a steel sample containing 10% B<sub>4</sub>C and Ni gradually shifted from austenite to lower temperature phases as the sample cooled. The change to pearlite and ferrite occurred in the mostly ferritic microstructures of steels (without cladding of B<sub>4</sub>C and Ni). The transformation from predominantly ferritic





Fig. 4. Photograph of the microwave-clad samples.

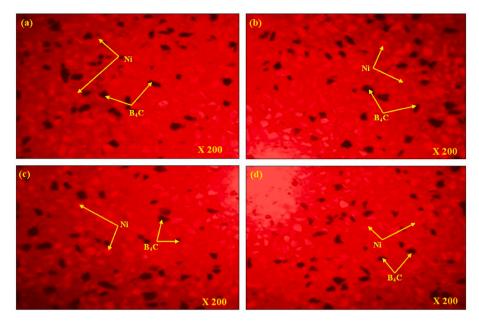


Fig. 5. Optical microscopic images of the cladding surface.

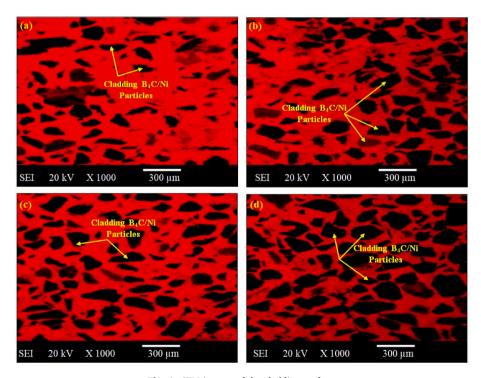


Fig. 6. SEM images of the cladding surface.

microstructures to pearlite and ferrite in steels, without the addition of  $B_4C$  (boron carbide) and Ni (nickel) cladding, signified a crucial metallurgical alteration. This transition was achieved due to the effect of heat treatment occurred during the processing, which involved heating and cooling the steel. It resulted in a more balanced microstructure. The presence of pearlite, a lamellar mixture of cementite and ferrite, enhanced hardness and strength, while ferrite contributed to ductility [59,60]. Since interlamellar spacing and the pearlite did not contribute noticeably to the strength of cladding compositions, ferrite grain size was the most essential parameter of concern in this study following microwave cladding of 10%  $B_4C$  and Ni on steel. There was evidence that during the solidification phase following microwave processing,

ferrite grains of cladding samples nucleated near austenite grain boundaries. It has been noted, however, that the residual strain and the ultimate austenite grain size influenced the ferrite grain size of coated steel. The retained strain and the ultimate austenite grain size were both affected by environmental factors such as cooling rate, composition, and deformation history. After undergoing microwave processing, the austenite grains in the cladding samples grew larger to reach their ultimate, completely recrystallized size [61,62]. The cladding surface and layer were seen in detail in the SEM picture shown in Fig. 7. There were fewer black pixels in the homogeneous cladding layer, as seen in Fig. 7(a & b) (yields a higher homogeneity). An even layer of coating of Ni and  $B_4C$  ensured that the substrate was adequately protected [63–65]. The

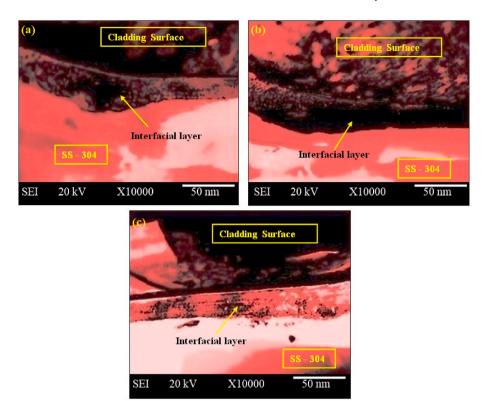


Fig. 7. SEM image of cladding layer and cladding surface.

uniform cladding layer ensured that there were no weak spots or gaps that could expose the substrate to harmful elements [66,67]. Uniform cladding surfaces increased the surface area available for adhesion, improving the bonding strength of the coating [68]. This helped to increase the lifespan of the coated product [56,57]. A uniform and consistent cladding layer enhanced the appearance of the coated surface. It provided a smooth and even surface that added to the overall visual appeal of the product [69–71].

The grain count of SS-304 alloy microwave cladding with a  $10\%~B_4C$  and Ni powder blend, was determined per square inch at  $500\times$  magnification. This analysis provided insights into the microstructural changes induced by the cladding process, offering valuable information about material properties and performance [55,72,73]. The number of grains (n) was observed by following Equation (1):

$$n = 2^{(G-1)} (1)$$

Where, n =the number of grains, G =grain size number ASTM.

$$G = [-6.644\log_{10}(l_a)] - 3.28 \tag{2}$$

Where,  $l_{\alpha} = \frac{V_{\nu\alpha}(L_T)}{N_{\alpha}}$ .

Here,  $V_{\nu\alpha}=$  fraction of  $\alpha$  phase (volume),  $L_T=$  length/magnification (test line),  $N_{\alpha}=$  grains intercepted (number).

$$l_{\alpha} = \frac{0.47 \times (493/500)}{95}$$

$$l_{\alpha} = 0.0048$$

Now again G from Equation (2),

$$G = [-6.644\log_{10}(0.0048)] - 3.28$$

G = 12.125.

Now from Equation (1)

$$n = 2^{(12.125-1)}$$

$$n = 2^{(11.125)}$$

$$n = 2233.35$$

In the examination of SS-304 alloy after microwave cladding with a 10%  $B_4C$  and Ni powder blend, a significant finding emerged: approximately 2233.35 grains were observed per square inch at  $500\times$  magnification. This numerical insight sheds light on the microstructural transformations resulting from the cladding process [55,74–76]. The rise in grain count signified that the microwave cladding had induced a finer grain structure in the material, which often led to enhanced mechanical characteristics, such as enhanced hardness and wear resistance [77–79].

# 3.2. XRD behavior

The XRD of a 10% B<sub>4</sub>C and Ni powder-clad SS-304 alloy is shown in Fig. 8. X-ray diffraction analysis of a cladding surface reveals the presence of  $Cr_2C$  (38.94°, 48.46°, 51.82°, 66.68°, 67.92°, 74.76°),  $Cr_{23}C_6$  $(34.3^{\circ}, 42.92^{\circ}), Cr_7BC_4 (29.32^{\circ}), Fe_5C_2 (25.94^{\circ}, 39.06^{\circ}, 78.96^{\circ}), and$ Fe<sub>23</sub>B<sub>6</sub> (45.7°) phases. After being clad with a combination of 10% B<sub>4</sub>C and Ni powder, SS-304 alloy gained increased hardness due to the formation of hard and carbide phases such as Cr<sub>2</sub>C, Cr<sub>23</sub>C<sub>6</sub>, and Cr<sub>7</sub>BC<sub>4</sub>. Cr<sub>2</sub>C (carbide phase) was formed due to the presence of chromium in the steel [80-82]. It provided excellent wear resistance [83,84]. However, excessive carbide formation led to brittleness and reduced the ductility of steel. Cr<sub>23</sub>C<sub>6</sub> was another chromium carbide phase that formed at high temperatures [85-87]. It had a similar effect on steel as Cr<sub>2</sub>C, but it was more stable and could provide better corrosion resistance. Cr7BC4 (boride phase) was formed by the addition of boron to steel [88-90]. It improved the hardness and wear resistance of steel. The Fe<sub>5</sub>C<sub>2</sub> carbide phase was formed by the reaction between iron and carbon [91–94]. It provided hardness and wear resistance to steel. The Fe<sub>23</sub>B<sub>6</sub> boride phase

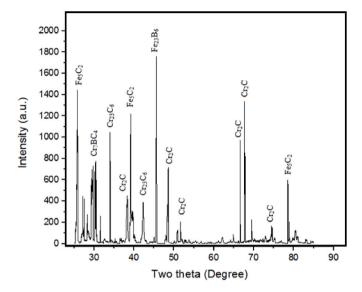


Fig. 8. XRD of SS-304 alloy with the microwave cladding of the mixture of 10 %  $B_4 C$  and Ni powder.

was formed by the addition of boron to steel. It improved the strength and ductility of steel [95–97].

#### 3.3. Surface hardness

The hardness of SS-304 alloy, after microwave cladding with a 10% B4C and Ni powder blend, plays a pivotal role in defining its surface characteristics. The standard deviation of Vickers hardness readings was within 1-2 HV units in the present study. The SS-304 alloy, originally possessing an average hardness of 210 HV, underwent a transformative change after microwave cladding with a Ni and 10% B<sub>4</sub>C powder blend. The resulting cladding surface exhibited a remarkable hardness of approximately 301 HV. This substantial increase of 43.33% in hardness showed a significant enhancement in material properties achieved through the cladding process [98–100]. The elevated hardness indicated improved resistance to wear, abrasion, and deformation, making SS-304 alloy with Ni and 10% B<sub>4</sub>C an ideal choice for applications demanding heightened surface durability and performance, such as in the aerospace and manufacturing industries [101-103]. The formation of carbide phases like Cr<sub>2</sub>C, Cr<sub>23</sub>C<sub>6</sub>, and Cr<sub>7</sub>BC<sub>4</sub> during microwave cladding played a pivotal role in boosting material hardness. These carbides contributed to the improved surface properties, providing superior wear resistance and increased durability [104-106]. The generation of these hard phases on the surface contributed to an increase in the resistance of the material to indentation [56,57], hence elevating the hardness of the cladding surface. In Fig. 9, hardness distributions for SS-304, the cladding layer, and the cladding surface were examined. Interestingly, the cladding layer and the cladding surface displayed uniform hardness distributions, indicative of consistent material properties [107-109]. Conversely, SS-304 exhibited significant hardness variability after solidification, suggesting irregularities in its microstructure [58,59]. Notably, the cladding layer exhibited slightly lower hardness when compared to the cladding surface [110,111]. This differential hardness profile reflected the influence of the microwave cladding process, which imparted distinct characteristics to each region.

# 3.4. Wear behavior

A wear test on the cladding surface of SS-304 with a coating mixture of Ni and 10% B<sub>4</sub>C was conducted to assess its performance and durability. In the wear test, a pin-on-disc machine operated at a sliding speed of 2 m/s, traversing a distance of 1000 m while applying an axial load of

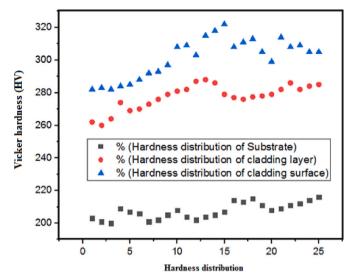


Fig. 9. Hardness distribution from SS-304 to cladding surface.

5 N. The cladding surface, comprising Ni and 10% B4C, displayed impressive performance attributes, featuring a low wear rate of 0.00308 mm<sup>3</sup>/m and a coefficient of friction measuring 0.1981. These results signified its exceptional wear resistance and frictional properties [112–114]. The exceptional wear resistance of the SS 304 cladding can be primarily attributed to the strategic incorporation of B<sub>4</sub>C with Ni powder during the cladding process. The addition of B<sub>4</sub>C, known for its remarkable hardness and abrasive properties, into the Ni matrix created a composite material that was significantly harder and more wear-resistant than the base SS 304 alloy [115-117]. When SS-304 was coated with the mixture of Ni and 10% B<sub>4</sub>C subjected to wear test, the B<sub>4</sub>C particles acted as abrasive reinforcements, reducing material loss and minimizing wear [118,119]. Additionally, the composite's microstructure, including fine grains and uniform distribution of B<sub>4</sub>C, enhanced its ability to withstand friction and stress [120,121]. This synergistic effect between the constituents enhanced the material's ability to withstand friction, abrasion, and wear [2]. However, the boron carbide (B<sub>4</sub>C) was indeed a crucial component known for enhancing a material's resistance to wear. However, sometimes in the context of dynamic stress, certain challenges arise. Cavitation erosion occurred at the interface between the cladding and the SS-304 substrate, leading to the degradation of the Ni-and-10%- $B_4C$  cladding [122-124]. The degradation mechanisms involved multiple factors. Columnar grains within the cladding sheared, leading to damage. Additionally, plastic deformation occurred within the SS-304 substrate due to the stress, further contributing to wear. Undulations and cyclic strain in the substrate led to cladding delamination [125-127]. The elastic nature of the cladding and transient loading caused it to separate from the substrate. In cases of cladding delamination due to cyclic dynamic loads, shear fractures propagated across the entire cladding thickness, resulting in cladding loss [55,58]. Furthermore, cladding undulations induced tensile stresses within the coating, further exacerbating the issue. These stresses initiated and propagated cracks, ultimately impacting the material's overall integrity [128–130]. The microwave cladding of a 10% B<sub>4</sub>C and Ni powder blend not only enhanced the wear resistance of SS-304 but also introduced phases like Cr<sub>2</sub>C, Cr<sub>23</sub>C<sub>6</sub>, and Cr<sub>7</sub>BC<sub>4</sub>. These newly formed carbide phases significantly contributed to the improved wear behavior of the material [131-133].

In this study, the cladding of Ni and  $10\%~B_4C$  on SS 304 induced grain refinement, leading to smaller grain sizes. Fine-grained materials had more grain boundaries, which acted as barriers to dislocation movement during wear [134–136]. This helped distribute stress and reduced the propensity for plastic deformation and wear. During wear, particles and wear debris were generated at the contact interface

[137–139]. The incorporation of  $B_4C$  in the cladding introduced hard particles with a solid lubricating effect. These particles reduced friction and abrasive wear by forming a protective layer on the contact surfaces [140,141]. The hard  $B_4C$  particles withstood abrasive forces, helping to minimize wear. Adhesive wear occurs when surfaces bonded and then separated, leading to material transfer between contacting surfaces [142–144]. The hardness and lubricating effect of the Ni and  $B_4C$  coating reduced adhesion by inhibiting material transfer. This helped prevent excessive wear and damage to the underlying material. Grooves were formed on the surface due to abrasive wear. The presence of hard  $B_4C$  particles contributed to groove formation as they abraded the surface (Fig. 10).

The wear mechanism and microstructural changes induced by cladding SS 304 with Ni and 10% B<sub>4</sub>C were of paramount importance in understanding the material's performance. The hard B<sub>4</sub>C particles dispersed within the Ni matrix substantially enhance wear resistance [145,146]. During wear, these hard particles acted as abrasion-resistant reinforcements, effectively reducing material loss due to abrasive contact. Additionally, the cladding process induced microstructural alterations such as grain refinement, dispersion of B<sub>4</sub>C, and possibly phase transformations, which collectively contributed to improved wear resistance [147-149]. The wear surface morphology of SS 304 steel coating with Ni and B<sub>4</sub>C showed a uniform and dense microstructure (Fig. 10). The presence of boron carbide particles in the coating enhanced the hardness and wear resistance of the material. The surface appeared to be smooth and without any visible cracks or discontinuities, indicating good adhesion and cohesion of the coating. Under high magnification, the surface revealed a uniform distribution of boron carbide particles within the nickel matrix. The particles were well dispersed and did not show any clustering, indicating good mixing and homogeneity. The surface also displayed a high degree of compaction and low porosity, indicating good coating quality [150,151]. The wear surface of the coating after tribological testing showed minimal wear, indicating excellent abrasion resistance (Fig. 10). The surface appeared to be free of any major wear marks, indicating good durability and performance [152]. The wear pattern was also uniform across the surface, indicating that the coating was performing consistently [153,154].

#### 3.5. Corrosion behavior

The corrosion test for the stainless steel-316 has been conducted with 3.5 wt Percent sodium chloride over 120 h. Corrosion weight loss for corroded SS-304 with coating the mixture of Ni and  $10\%~B_4C$  was found

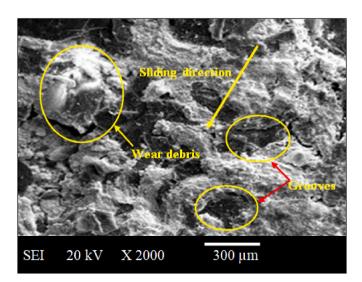


Fig. 10. Wear surface morphology of SS-304 with coating the mixture of Ni and  $10\%~B_4C$ .

to be 0.42 mg. A Tafel polarization study was also conducted on SS-304 coated with a Ni and 10% B<sub>4</sub>C mixture. In this investigation, polished specimens of SS-304 with this coating were immersed in a corrosion medium with varying concentrations of NaOH (0.05 M and 1 M) at an elevated temperature of 300 °C. A stable open circuit potential (OCP) was established. Subsequently, potentiodynamic current-potential curves were generated by polarizing the specimen cathodically to -250 mV and anodically to +250 mV concerning the OCP, with a scan rate of 1 mV s-1. This study aimed to evaluate the material's corrosion behavior and resistance under different conditions, providing insights into its durability and suitability for specific environments. In Fig. 11, potentiodynamic polarization curves for the SS-304 coated with a Ni and 10% B<sub>4</sub>C mixture immersed in a NaOH solution at an elevated temperature of 300 °C are depicted. This electrochemical analysis yielded crucial parameters to assess the material's corrosion behavior. Among these parameters were the corrosion rate, cathodic Tafel slope, anodic Tafel slope, corrosion potential, corrosion current density, and corrosion potential. The results revealed a notable correlation between the concentration of NaOH and the corrosion current density. As the NaOH concentration increased, there was a corresponding rise in the corrosion current density [153,154]. This observation suggested that the presence of a more concentrated NaOH solution accelerated the corrosion process of the SS-304 material with the Ni and 10% B<sub>4</sub>C coating. Fig. 12 shows the surface morphology of corroded SS-304 with coating the mixture of Ni and 10% B<sub>4</sub>C and in which few dark spots may be found. This was because the Ni and B<sub>4</sub>C in the coating acted as corrosion inhibitors that provided additional protection to the steel against corrosion. In the presence of a corrosive agent, the Ni and B<sub>4</sub>C reacted with the corroding species and formed a protective barrier on the steel surface [153,154]. The morphology of the corrosion varied from simple pitting to more complex forms of corrosion such as crevice corrosion, intergranular corrosion, and selective attack of specific phases [55,58].

However, the corrosion resistance of SS 304 after coating with Ni and  $B_4 C$ , while improved, faces several limitations and areas for enhancement. Pitting corrosion remains a concern in aggressive environments, necessitating strategies to enhance pitting resistance. Ensuring strong coating-substrate adhesion and minimizing microscopic porosity are critical for long-term stability. Specific corrosion agents require tailored coatings. Thermal cycling can stress the coating, requiring improved thermal stability [153,154]. Addressing cost and scalability concerns is essential for practical industrial adoption. Field testing under real conditions is needed to validate performance, and regulatory compliance must be met. Overcoming these limitations will lead to more robust and versatile corrosion-resistant coatings for SS 304.

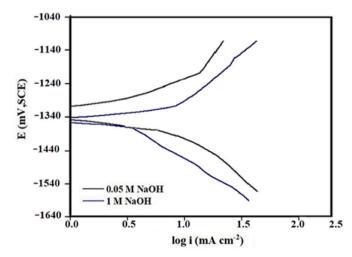


Fig. 11. The Tafel plots for corroded SS-304 with coating the mixture of Ni and 10%  $B_4C$  in different concentrations of NaOH at 300  $^{\circ}C$ .

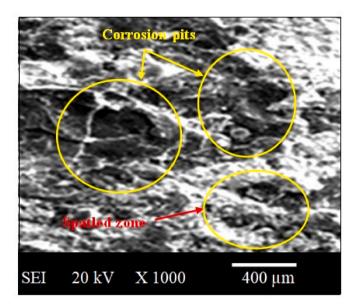


Fig. 12. Surface morphology of corroded SS-304 with coating the mixture of Ni and 10%  $B_4 C$ .

# 3.6. Comparison of present study work with previous literature

The present study reported a significant 43.33% enhancement in hardness. This result suggested that the Ni and 10% B<sub>4</sub>C coating applied using microwave hybrid heating has effectively improved the hardness of SS-304. Fangyong Niu et al. [33] showed that the microhardness of the TiC ceramic coating on 304 stainless steels reached 654.7  $\pm$  41.0 HV0.2, which was 3.15 times higher compared to WAAM-304. This suggested that the TiC coating significantly enhanced the hardness of the substrate. Abhijit Pattnayak et al. [34] showed that the nano hardness of the Al<sub>2</sub>O<sub>3</sub>-CeO<sub>2</sub>-rGO coating was approximately 87% higher than that of Al<sub>2</sub>O3–CeO<sub>2</sub> coating. This indicated that the addition of Ceria and reduced graphene oxide significantly enhanced the hardness of the coating. The study of Zhaoliang Li et al. [35] found that when the amount of independent Al<sub>2</sub>O<sub>3</sub> was increased to 30%, a strong metallurgical bond developed. This result suggested that the composition of the coating is critical for achieving a strong bond. Shuwei Guo et al. [36] showed that the TiO2 coating exhibited better protective effects on reducing 316 SS. This indicated that TiO<sub>2</sub> provided improved protection against corrosion in the given environment. Table 3 shows the summary of the comparison of the present study work with previous literature.

#### 4. Conclusions

The microwave cladding technique for depositing Ni and B<sub>4</sub>C coatings on 304 stainless steels has yielded significant advancements in various aspects. The key findings and improvements include a remarkable 43.33% enhancement in material hardness, indicative of improved wear resistance and mechanical properties. The uniform distribution of B<sub>4</sub>C and Ni on the cladding surface ensured consistent material performance. Microstructural analysis revealed the presence of approximately 2233.35 grains per square inch at 500× magnifications, highlighting the refined grain structure achieved during the cladding process. Wear testing demonstrated a low wear rate of 0.00308 mm<sup>3</sup>/m and a favorable coefficient of friction measuring 0.1981, underlining the material's suitability for applications with demanding frictional conditions. Moreover, the corrosion behavior was assessed, revealing a corrosion weight loss of only 0.42 mg for corroded SS-304 with the Ni and 10% B4C coating, indicating notable corrosion resistance. The presence of various carbide phases, including Cr<sub>2</sub>C, Cr<sub>23</sub>C<sub>6</sub>, Cr<sub>7</sub>BC<sub>4</sub>, Fe<sub>5</sub>C<sub>2</sub>, and Fe<sub>23</sub>B<sub>6</sub>, within the cladding further contributed to the material's enhanced mechanical and wear properties. Limitations and uncertainties in the findings include potential variations in real-world application conditions, long-term durability, and the need for further environmental testing. Additionally, while microwave cladding shows promise, specific factors affecting material behavior may require more comprehensive investigation for practical industrial implementations. Exploring the future scope for the present study can lead to significant advancements in materials science and engineering. Here are some suggestions for future research and development as In-depth optimization of microwave parameters, such as power, frequency, and exposure time, to further enhance the quality and performance of the coatings; investigation of the thermal behavior of the coatings under various conditions; Investigate the potential benefits of multi-material or multilayer coatings that combine Ni, B4C, and other materials to achieve enhanced properties; Explore the development of nanocomposite coatings. The combination of SS304 with Ni and B<sub>4</sub>C coating can be used in the aerospace industry such as including aircraft structural parts and fuel system components. It can be used in the construction of radiation shielding enclosures and containers in nuclear power plants and medical facilities. Coated SS 304 steel with Ni and B<sub>4</sub>C may be used in exhaust systems, brackets, and fasteners. It can be also used in boat components, underwater pipelines, and marine hardware.

# **Ethical approval**

Not applicable.

**Table 3**Summary of comparison of present study work with previous literature.

S. No.	Ref. No.	Substrate/ Coating Plate	Cladding/Coating Particles	Experimental Set-up	Properties
1	Present Study Work	SS-304	Ni and 10% B <sub>4</sub> C	Microwave Hybrid Heating	43.33% enhancement in hardness
2	Fangyong Niu et al. [33]	304 stainless steel	TiC ceramic coating	Wire Arc Additive Manufacturing (WAAM)-Laser Cladding (LC) hybrid	The microhardness of the coating reaches 654.7 $\pm$ 41.0 HV $_{0.2}$ , which is 3.15 times higher compared to WAAM-304
3	Abhijit Pattnayak et al. [34]	17-4 PH steel	Alumina (Al2O3) doped with 0.8% Ceria (CeO2) and reduced graphene oxide (rGO, 0–0.2%)	High-Velocity Oxygen Fuel (HVOF) thermal spray process	Nano hardness of Al $_2$ O $_3$ –CeO $_2$ -rGO coating is $\approx$ 87% higher than that of Al $_2$ O $_3$ –CeO $_2$ coating
4	Zhaoliang Li et al. [35]	Carbon steel	Al <sub>2</sub> O <sub>3</sub> –TiB <sub>2</sub> –TiC ceramic coatings	Laser cladding	When the amount of independent ${\rm Al}_2{\rm O}_3$ was increased to 30%, a strong metallurgical bond developed
5	Shuwei Guo et al. [36]	316 stainless steel	$\rm ZrO_2$ and $\rm TiO_2$ ceramic coatings	Air plasma spraying and then exposed to oxygenated sub- and supercritical water	The $TiO_2$ coating exhibited better protective effects on reducing 316 SS

#### Consent to participate

Not applicable.

#### Consent to publish

All authors have read and approved this manuscript.

#### Author contributions

Conceptualization, SS, SPD, APS, VAS, KAM; methodology, SS, SPD, APS, VAS, KAM; formal analysis, SS, SPD, APS, VAS, KAM; investigation, SS, SPD, APS, VAS, KAM; writing—original draft preparation, SS, SPD, APS, VAS, KAM; writing—review and editing, SS, AK, MIK, MA, EMTE; supervision, SS, AK, MIK, MA, EMTE; project administration, SS, AK, MIK, MA, EMTE; funding acquisition, SS, AK, MIK, MA, EMTE. All authors have read and agreed to the published version of the manuscript.

#### Data availability statement

The data that support the findings of this study are available within the manuscript.

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#### Conflict of interests/competing interests

The authors declare no competing interests.

#### **Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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