SYSTEM Cu₃In₅S₉ – CuFeIn₃S₆ Bakhtiyarly I.B., Kurbanova R.J., Abdullayeva Sh.S. Institute of Catalysis and Inorganic Chemistry ANAS Az-1143 Baku, H. Javid ave. 113

Phase equilibria in the Cu₂S-In₂S₃-FeS quasy ternary system were studied along the Cu₃In₅S₉-CuFeIn₃S₆ section by means of differential-termal analysis, X-ray diffraction, microstructure and density determination technique and its phase diagram was constructed (see Figure). Cu₃In₅S₉ and CuFeIn₃S₆, which we take as the starting component are congruent melting compounds formed in quasibinary systems of Cu₂S-In₂S₃ and FeIn₂S₄-CuInS₂. It was found that the Cu₃In₅S₉-CuFeIn₃S₆ section is quasibinary and its phase diagram refers to an eutectic type. Liquidus of this system consists of two branches intersecting eutectic point which characterize non-equilibrium L $\leftrightarrow \sigma_1(Cu_3In_5S_9)+\delta(CuFeIn_3S_6)$ and indicating primary crysallization of the solid solutions based on $\sigma_1(Cu_3In_5S_9)$, $\delta(CuFeIn_3S_6)$. The coordinate of the eutectic point is 55 mol% CuFeIn₃S₆ and crystallizes at 1200K. The composition of the eutectic point was confirmed by the construction of the Tamman triangle.

Based on the results of the microstructure analysis conducted on these samples the boundaries of the solid solution regions were clarified.



Phase diagram of the $Cu_3In_5S_9 - CuFeIn_3S_6$ system

Thus, the phase diagram of the $Cu_3In_5S_9 - CuFeIn_3S_6$ section was constructed based on the complex results obtained from the mentioned methods of physical and chemical analysis. The diagram is eutectic type and at room temperature (300 K) region of solid solutions based on $Cu_3In_5S_9$ reaches 5 mol%, on $CuFeIn_3S_6$ it reaches 2 mol%. The coordinate of the eutectic point 55 mol% $CuFeIn_3S_6$ at 1200 K.